

## Chapter-2 Homework

$$\begin{aligned} E &= m \gamma c^2 & \beta &= v/c \\ E &= KE + m c^2 & \gamma &= 1/[1-\beta^2]^{1/2} \\ E &= E^2 = p^2 c^2 + m^2 c^4 & 1 \text{ u} &= 931. \text{ MeV} \\ p &= m \gamma v = m \gamma \beta c \end{aligned}$$

#1 (a)  $\gamma = 1.0005$   $p = m \gamma v = 1.673 \times 10^{-27} \text{ Kg} (1.00005) (3.0 \times 10^6 \text{ m/s}) = \underline{5.02 \times 10^{-21} \text{ Kg m/s}}$   
 (b)  $\gamma = 1.1550$   $p = m \gamma v = 1.673 \times 10^{-27} \text{ Kg} (1.155) (1.5 \times 10^6 \text{ m/s}) = \underline{2.89 \times 10^{-19} \text{ Kg m/s}}$   
 (c)  $\gamma = 2.2942$   $p = m \gamma v = 1.673 \times 10^{-27} \text{ Kg} (2.294) (2.7 \times 10^6 \text{ m/s}) = \underline{1.03 \times 10^{-18} \text{ Kg m/s}}$

(d)  $p = m \gamma \beta c = (939 \text{ MeV}/c^2) (1.00005) (.01) c = \underline{9.39 \text{ MeV}/c}$   
 (e)  $p = m \gamma \beta c = (939 \text{ MeV}/c^2) (1.155) (.5) c = \underline{540 \text{ MeV}/c}$   
 (f)  $p = m \gamma \beta c = (939 \text{ MeV}/c^2) (2.294) (.9) c = \underline{1930 \text{ MeV}/c}$

#8  $\beta = 0.95$   $\gamma = 3.20$   $E^2 = p^2 c^2 + m^2 c^4$   
 (a) Rest Energy  $p = 0$   $E = m c^2 = (938.3 \text{ MeV}/c^2) c^2 = \underline{938.3 \text{ MeV}}$   
 (b)  $E = m \gamma c^2 = (938.3 \text{ MeV}/c^2) (3.20) c^2 = \underline{3004 \text{ MeV}}$   
 (c)  $KE = E - m c^2 = 3004 \text{ MeV} - 938.3 \text{ MeV} = \underline{2067 \text{ MeV}}$

#9  $KE = 5 \text{ m} = 2555 \text{ KeV}$  with  $m = 511 \text{ KeV}/c^2$   
 (a)  $E = KE + m c^2 = \underline{3066 \text{ KeV}}$   
 (b)  $\gamma = E / m c^2 = 3066 \text{ KeV} / 511 \text{ KeV} = 6.0$   
 $\beta = [1 - (1/\gamma)^2]^{1/2} = 0.986$  or  $v = \underline{0.986 c}$

#11  $KE = 50 \text{ GeV}$   $E = KE + m c^2 = 50 \text{ GeV} + 0.939 \text{ GeV} = 50.939 \text{ GeV}$   
 (a)  $p c = \sqrt{E^2 - m^2 c^4} = 50.930 \text{ GeV}$  or  $p = \underline{50.930 \text{ GeV}/c}$   
 (b)  $\beta / c = p/E = (50.930 \text{ GeV}/c) / (50.939 \text{ GeV}) = 0.99982$   $v = \underline{0.99982 c}$

#18  $^{55}\text{Cr}_{24} \rightarrow ^{55}\text{Mn}_{25} + ^0\text{e}_1$   $A \rightarrow B + e$   $M_A \sim M_B \gg m_e$   
 (a)  $\Delta M = M_A - M_B = 54.928 \text{ u} - 54.9244 \text{ u} = .0035 \text{ u} = \underline{3.26 \text{ MeV}/c^2}$

The decay  $Q = (M_A - M_B - m_e) c^2 = 0.00295 \text{ u} = 2.75 \text{ MeV}$   
 (Not exactly what was asked for, BUT!!)

(b) After the decay the Mn nucleus remains virtually stationary ( $T_{\text{Mn}} \sim 0$ ) and the net decay energy goes to the electron,

$M_A c^2 = M_B c^2 + T_B(\sim 0) + m_e c^2 + T_e$  *conservation of energy*  
 $T_e \cong \Delta M c^2 - m_e c^2 = 3.26 \text{ MeV} - 0.511 \text{ MeV} = \underline{2.75 \text{ MeV}}$

OR

$T_e = (M_A c^2 - M_B c^2 - m_e c^2) \cong Q = 2.75 \text{ MeV}$

